Marathon Petroleum Company un	SAFE WORK INSTRUCTION				HS-SWI-019
SALT LAKE REFINERY	Acid Caustic Handling Page 1 of 9				Page 1 of 9
RESPONSIBLE DEPT.	CONTENT STEWARD APPROVED BY				
Safety	Industrial Hygienist			ESS Manager	
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CONTENTS

1.0	Intro	duction	.2
	1.1	Purpose	.2
	1.2	Scope	.2
	1.3	Corporate References	.2
2.0	Defin	itions	.2
3.0	Roles	and Responsibilities	3
			_
	3.1	Safety Department	.3
	3.1 3.2	Safety Department Owning Department Supervision	.3 .3
	3.1 3.2 3.3	Safety Department Owning Department Supervision Owning Department Personnel	.3 .3 .4
	3.1 3.2 3.3 3.4	Safety Department Owning Department Supervision Owning Department Personnel Servicing Group	.3 .3 .4 .4
	3.1 3.2 3.3 3.4 3.5	Safety Department Owning Department Supervision Owning Department Personnel Servicing Group Contractors	.3 .3 .4 .4 .4

LIST OF TABLES

4.0	Practi	ices	.4
	4.1	General Safe Practices	5
	4.2	Laboratory Safe Practices	5
	4.3	Acid Release to Sewer Precautions	5
	4.4	Uncontrolled release of acid to sewers	6
	4.5	Goggle Required Areas	6
	4.6	PPE Requirements	6
	4.7	Safety Showers	8
	4.8	Chemical Characteristics	8
5.0	Revie	w and Revision History	.9
	5.1	Procedure Review	9
	5.2	History of Revisions	9

Table 1	Terms and Definitions	2
Table 2	Revision History	9

Marathon Petroleum Company P

SALT LAKE REFINERY

SAFE WORK INSTRUCTION

Acid Caustic Handling

1.0 INTRODUCTION	ON	
1.1 Purpose	1.1.1	This Safe Work Instruction outlines specific steps to be taken by employees and contractors to minimize potential exposure to acid, caustic and ammonia services at the Salt Lake Refinery.
1.2 Scope	1.2.1	This Safe Work Instruction applies to both employees and contractors.
	1.2.2	This safe work instruction provides specific information in the following areas:
		Engineering controls, administrative controls, procedures, safe handling, and personal protective equipment when working on or operating equipment in acid, caustic or ammonia service.
		Will also provide information on acid release precautions, respiratory protection, and medical treatment when a person comes in contact with or inhale acids or caustics.
1.3 Corporate References	The follo	owing sections describe references used to generate this Safe Work on.
	1.3.1	Marathon Standards, Policies and Procedures
		Corporate: HLT-2001 Industrial Hygiene Program
		Local: HS-SWI-001 Safe Work Permit
		Local: Respiratory Protection Program
	1.3.2	Government Regulations

2.0 **DEFINITIONS**

The following terms and definitions are used in this document.

Term	Definition
Ammonia (NH₃)	Colorless / Flammable gas with a characteristic pungent smell that can cause tissue burns and is a strong respiratory irritant.
НЕРА	High Efficiency Particulate Air
HMIS	Hazardous Materials Identification System
Hydrogen Chloride	Non-flammable, toxic gas. At room temperature, it is a colorless gas, which forms white fumes of hydrochloric acid upon contact with moisture, including skin moisture.
IDLH	"Any condition that poses an immediate or delayed threat to life (IDLH) or that would cause irreversible adverse health effects or that would interfere with an individual's ability to escape unaided from a permit space".
MoC	Management of Change

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Table 1 Terms and Definitio	ns
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Term	Definition	
PEL	The permissible exposure limit (PEL or OSHA PEL) is a legal limit in the United States for exposure of an employee to a chemical substance or physical agent.	
рН	measure of the acidity or basicity of an aqueous solution	
PPE	Personal Protective Equipment	
PVC	Polyvinyl Chloride	
SCBA	Self-Contained Breathing Apparatus (Fresh Air)	
Sodium Hydroxide (NaOH)	Also known as lye or caustic soda, is a caustic metallic base.	
Sulfur Dioxide (SO ₂)	A colorless, nonflammable, water-soluble, suffocating gas, SO2, formed when sulfur burns: used chiefly in the manufacture of chemicals such as sulfuric acid	
Sulfuric Acid (H ₂ SO ₄)	Colorless (pure) to dark brown, oily, dense liquid with a specific gravity almost double that of water.	

3.0 ROLES AND RESPONSIBILITIES

3.1 Safety	Health and Safety is responsible for:		
Department	 Maintaining this instruction with a program review at least every five (5) years 		
	 Providing training material to support the training responsibilities of Operations and the Training Department 		
	Implementing the EXAM process to anticipate and recognize potential exposures.		
	 Provide reports to management on any compliance issues 		
3.2 Owning	Supervision is responsible for the following:		
Department Supervision	ensuring this policy is implemented consistently from shift to shift and across the refinery		
-	providing resource materials to safely handle acids caustics and ammonia		
	 performing audits to ensure compliance 		
	reviewing this policy on a pre-determined frequency.		
	Ensure that those under their supervision are properly trained in the hazards of acids and caustics, safe work procedures, and the proper use of PPE.		
	Ensure that the MOC process is followed when unit conditions require the use of temporary hard piping in place of utility hoses.		

Marathon Petroleum Company ur	SAFE WORK INSTRUCTION	HS-SWI-019		
SALT LAKE REFINERY	Acid Caustic Handling	Page 4 of 9		
2 2 Owning	Owning Department is responsible for:			
Department Personnel	 following the guidelines of this policy and operating propreserve the integrity of acid and caustic systems. 	ocedures which		
	 determining which process streams contain acid and caustic along a general understanding of what concentrations may be present. 			
	the proper use and upkeep of PPE to ensure it is availadealing with acid or caustic releases or response to leases	able for use when aks.		
	ensuring all PPE and safety equipment (e.g. Acid Suits Nitrile Gloves, respiratory protection, etc.) are replaced unfit for use.	s, Face Shields, I if found to be		
3.4 Servicing	Servicing Group is responsible for:			
Group	following the guidelines of this policy while ensuring the caustic systems are maintained.	at acid and		
	e for use when eaks.			
	ensuring all PPE and safety equipment (e.g. Acid Suits Nitrile Gloves, respiratory protection, etc.) are replaced unfit for use.	s, Face Shields, I if found to be		
3.5 Contractors	Contractors are responsible for:			
	following the guidelines of this policy while ensuring the caustic systems are maintained.	at acid and		
	proper use and upkeep of PPE to ensure it is available dealing with acid and caustic releases or response to lease of the second sec	e for use when eaks.		
	ensuring all PPE and safety equipment (e.g. Acid Suits Nitrile Gloves, respiratory protection, etc.) are replaced unfit for use.	s, Face Shields, I if found to be		
3.6 Engineering	Engineering is responsible for:			
	ensuring that equipment containing acid or caustic are	closed systems.		
	designing control measures to prevent backing acid or caustic into other systems by insuring higher pressure, installation of check valves and block valves, and maintenance of these valves and provisions for testing.			

4.0 PRACTICES

The following section details the steps required to manage acid and caustic containing systems.

Marathon Petroleum Company P		SAFE WORK INSTRUCTION	HS-SWI-019
SALT LAKE REFINERY		Acid Caustic Handling	Page 5 of 9
4.1 General Safe Practices	4.1.1	Prior to opening lines or equipment which hav caustic, control of hazardous energy, line brea equipment opening, procedures must be follow	e contained acid or aking, and process wed to ensure that
		equipment is properly isolated, depressurized rinsed to a PH neutral condition.	, purged and
	4.1.2	Know the location and test the operability of safety showers and eyewash stations in the area where work is being performed.	
	4.1.3	Do not allow acids and caustics to come in co other.	ntact with each
	4.1.4	Have a large supply of water available when h caustic in the event of a spill	andling acid and
	4.1.5	Acids attack metal releasing hydrogen, a high	ly flammable gas.
	4.1.6	When opening railroad cars or lines containing ensure no source of ignition is present.	g sulfuric acid,
4.2 Laboratory Sat	fe 4.2.1	Work with chemicals in a lab should be done i	n a fume hood.
Practices	4.2.2	Where available, corrosives should be purcha with a protective plastic film so that splash is i container be dropped.	sed in containers nhibited should the
	4.2.3	Keep the container sizes and quantities on ha possible, consistent with the rate of use. Store itself to avoid cross contamination.	nd as small as e each class by
	4.2.4	Bottles containing corrosive materials from ref streams for lab analysis, should be properly la HMIS tag and capped.	inery process beled using a
	4.2.5	Do not allow corrosives to come in contact wit Sulfuric acid can be explosive on contact with	h organics. diesel fuels.
4.3 Acid Release t Sewer	o Acid car	 be released during process leaks, draining equipment. Prior to releasing acid to following must occur: 	uipment, or when the sewer, the
Precautions	4.3.1	Flush the Alky sewers with city or firewater pri acid to the sewer system.	or to introducing
	4.3.2	Inform the OMD board operator prior to releas sewer so they can alert the Traps operator of pH water being sent to the WWTF.	ing acid to the the incoming low
	4.3.3	Controlled acid draining should be kept to an a	absolute minimum.
	4.3.4	Never purposefully drain more than 10 gallons sewer. If draining more than 10 gallons is req HPDN Foreman. It will be necessary to devel safely dispose of acid.	s of acid to the uired contact the op a procedure to

Marathon Petroleum Company P	SAFE WORK INSTRUCTION HS-SWI-019		HS-SWI-019
SALT LAKE REFINERY	Acid Caustic Handling Page 6 of 9		Page 6 of 9
4.4 Uncontrolled release of acid sewers	4.4.1 I to	If an uncontrolled release greater than 10 gall sewer system or the ground call 4900 and foll reporting procedure found in the Salt Lake Cit Emergency Response Plan. All personnel sh from 268, 242, and 241 tank dikes.	ons enters the ow the emergency y Refinery ould be evacuated
4.5Goggle Requir Areas	ed 4.5.1	4.5.1 Areas around equipment and vessels that contain chemicals pose a significant acute hazard to the eyes should be permanently designated as a "Goggle Area". Goggles are required to be worn at all times when crossing or working wit the boundaries of the designated goggle area.	
	4.5.2	Painted yellow lines should designate the area with black lettering. The ground around the equipment should read at a minimum "GOGGLE AREA". The goggle area should be marked at least 6 feet from the equipment or vessel whenever the surface permits.	
	4.5.3	3 Visible signs alerting people to the designated goggles area als recommended in all cases and are required when the boundary cannot be painted. Where signs must be relied upon to alert personnel to the designated goggle area, such as areas where the ground cannot be painted (e.g., tank farm or unpaved areas the signs must be clearly visible at all pedestrian approaches or next to the equipment.	
	4.5.4	.4 An example of a sign is as follows;	
	C	AUTION	

CAUTIUN SAFETY GOGGLES REQUIRED IN THIS AREA

4.6 PPE

Requirements

The PPE requirements below are to be followed when conditions warrant.

- **4.6.1** Level D PPE is required when no respiratory protection is required, and minimal skin protection is needed such as conducting routine maintenance or sampling. Level D consists of:
 - a) Fire retardant coveralls (FRC) / work clothes
 - b) Work boots
 - c) Hard hats
 - d) Safety glasses
 - e) Goggles are required and must be immediately available

Marathon Petroleum Company u	SAFE WORK INSTRUCTION HS-SWI-01		HS-SWI-019
SALT LAKE REFINERY	Acid Caustic Handling		Page 7 of 9
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	4.6.2	Level D Prime PPE is for work requiring prote chemical splash and the vapor concentration Level D Prime PPE consists of:	ection against is below the PEL.
		a) FRC Work Boots	
		b) Goggles and face shield	
		c) PVC or nitrile chemical jacket	
		d) PVC or nitrile bibs or coverall	
		e) PVC or nitrile gloves.	
	4.6.3	Level C PPE is required when the vapor cond the permissible exposure limit (PEL) but does the PEL, there is a potential for splash hazard or caustic spills, containing small leaks or pad are small, controlled, and nonemergency. Leve of:	entration is above not exceed 50x d, cleaning up acid king failures which vel C PPE consists
		 Full face, air purifying respirator with com acid cartridge 	bination HEPA and
		b) One- or two-piece chemical splash suit (F	VC)
		c) Goggles/face shield combination	
		d) Chemical resistant boots or boot covers (neoprene or PVC)
	4.6.4	.4 Level B PPE is required when the vapor concentration may exceed 50X the PEL, a high potential for splash, containing la leaks or packing failures that are not controlled. Level B PPE consists of:	
		a) Pressure demand full-face SCBA	
		b) Chemical resistant clothing; hooded, one	, or two piece
		c) Chemical splash suit (PVC)	
		d) Chemical resistant gloves (PVC)	
		e) Chemical resistant boots or boot covers (neoprene or PVC)
	4.6.5	Level A PPE is required when vapor concentruk unknown, maximum skin protection is require uncontrolled and requires emergency respon- consists of:	ations are IDLH or d, or a release is se. Level A PPE
		a) Pressure demand full-face SCBA	
		b) Totally encapsulating chemical protective	suit (chemrel)
	4.6.6	Laboratory PPE requirements are as follows acids or caustics:	when handling
		 Safety glasses with side shields are required little or no potential for a splashing hazard 	ired when there is d.
		 b) Goggles or a face shield are required whe exists. 	en a splash hazard
		C) Laboratory apron (PVC) and gloves (PVC) manual handling of acids and caustics.) is required for all

Marathon Petroleum Company ur	SAFE WORK INSTRUCTION HS-SWI-019		HS-SWI-019
SALT LAKE REFINERY	Acid Caustic Handling Page 8 of 9		Page 8 of 9
4.7 Safety Showe	rs 4.7.1	At each point where acid or caustic is handled, there must be ready access to safety showers and eye wash stations. There must be a minimum of 15 minutes of warm water at a high enough volume to drench the user. Stations must be equipped with an automatic alarm system designed to summon help from adjacent areas. Safety showers must be supplied with 20 to 30 degree C tempered water to meet OSHA standards (min. 30 GPM flow), be within 50 feet of an unloading area, and be "winterized" (freeze protected).	
4.8 Chemical	4.8.1	Ammonia:	
Characteristic	S	a. Colorless / Flammable gas with a character that can cause tissue burns and is a strong	istic pungent smell respiratory irritant.
		 Exposure may cause severe injury to respir common 35% laboratory solution can cause burns. 	atory system and e severe skin
		c. High concentrations may cause temporary	blindness.
	4.8.2	Ammonia Exposure Limits:	
		a. Short Term Exposure Limit 35 ppm	
		b. 8-hour permissible exposure limit 25 ppm	
		c. IDLH 300 ppm	
	4.8.3	Sodium Hydroxide (NaOH):	
		a. Sodium Hydroxide is concentrated in a wate the facility. When dry, it is in the form of wh	er solution within ite pellets or flakes
		b. Also known as lye or caustic soda	
		C. Is considered a severe eye hazard	
		d. Will destroy tissue on contact in solid or in a solution	a concentrated
	4.8.4	Hydrogen Chloride (anhydrous)	
		a. Non-flammable, toxic gas by all routes of ex	xposure
		At room temperature, it is a colorless gas, v fumes of hydrochloric acid upon contact wit including skin moisture.	vhich forms white h moisture,
		 Inhalation of the fumes can cause coughing inflammation of the nose, throat, and upper and in severe cases, pulmonary edema, cir failure, and death. 	g, choking, respiratory tract, culatory system
		d. Skin contact can cause redness, pain, and	severe skin burns.
		e. Exposure above 1500 ppm, airborne conce fatal in minutes.	ntration, can be
	4.8.5	Sulfuric Acid (H2SO4) Characteristics	

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Marathon Petroleum Company u	SAFE WORK INSTRUCTION HS-SWI-019		
SALT LAKE REFINERY	Acid Caustic Handling Page 9 of 9		
	 Colorless (pure) to dark brown, oily, or specific gravity almost double that of 	lense liquid with a water.	
	 b. Corrosive and reactive properties ma hazardous when it comes in contact v such as sugars, fats, and proteins that 	ke it extremely with organic substances, at make up human tissue.	
	 Acids react violently with various stre generating heat and fumes. 	ngths of acids and alkalis	
	 Sulfuric acid mist is a human carcinog must be provided to keep concentrati limits. 	gen and good ventilation ons below exposure	

- e. Sulfuric acid will also react violently with metal powders, carbides, chlorates, fulminates, nitrates, picrates and strong oxidizing, reducing or combustible organic materials.
- f. Contact with acid can cause severe, deep burns to tissue.

5.0 REVIEW AND REVISION HISTORY

5.1 Procedure	The Safe Work Instruction will be reviewed every 5 years.
Review	

5.2 History of Revisions Table 2 provides the revision history for this Safe Work Instruction.

Table 2 Revision History

Revision	Date	Change Author	Reason for Change
1.0	12/13/1983		Original Issue
2.0	12/03/2012		Revision
2.1	01/06/2016		Updated to New SWI Format
3.0	1/31/2021	J.Moffitt	Updated to Marathon SWI Format